

2020 Michigan Winter Operations Conference

Day 3 – Thursday, October 15th

Environmental Impacts of Deicers



Laura Fay

2020 Michigan Winter Operations Conference

Oct. 15, 2020



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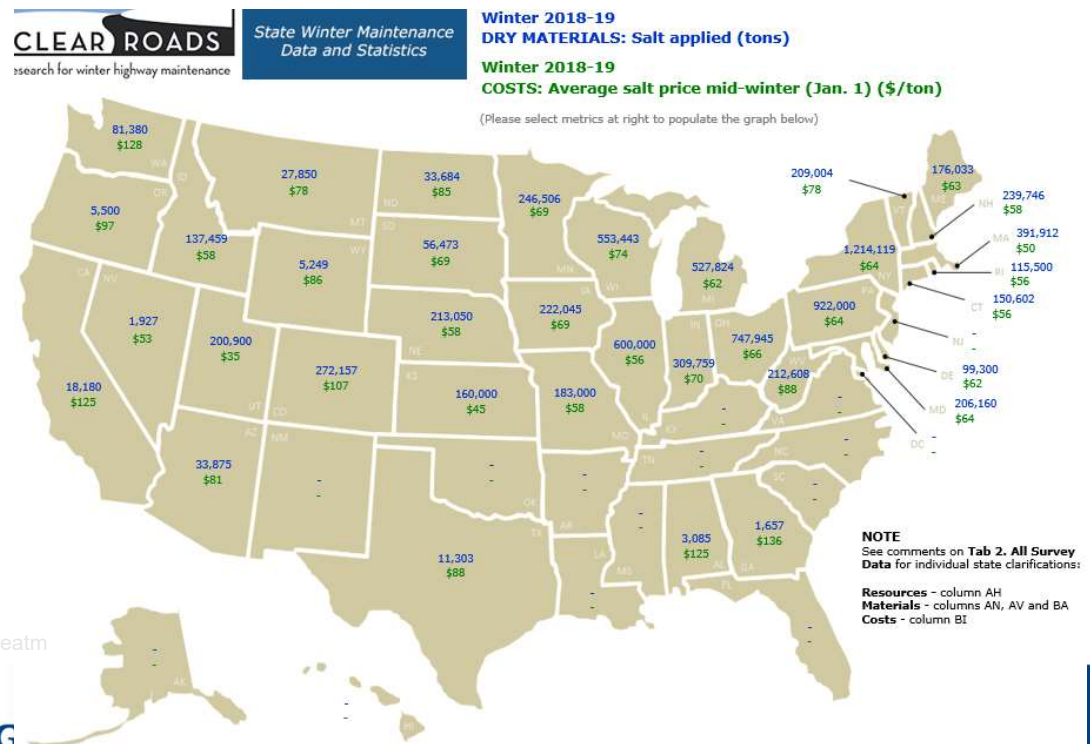
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Why Salt?

1. It is fairly cheap
2. It works
3. It has been proven to create safer road conditions!



https://www.northcountrypublicradio.org/news/images/Winter_road_treatment_using_salt_brine.jpg



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What impacts do we see?

- Corrosion

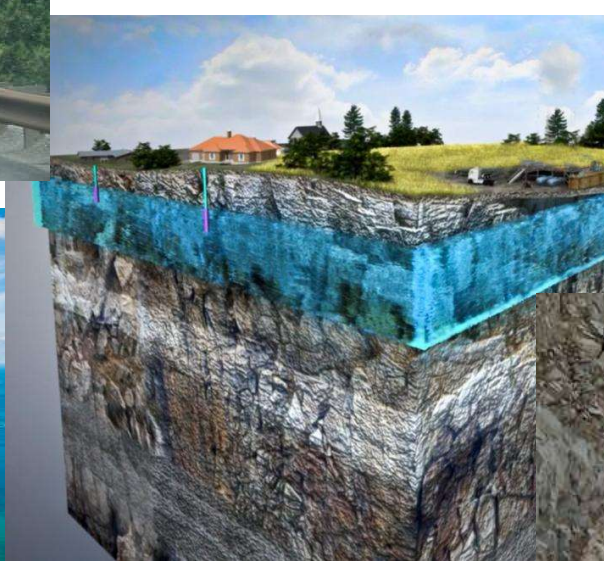


- Scaling



- Damage to rebar

What impacts can NOT see?



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Clear Roads QPL

(formerly known as PNS QPL)

Qualified Products List | Clear Roads

clearroads.org/qualified-product-list/

Apps New Tab MUS Wellness - Ho... Welcome to BioRAF... wtimesheets.coe... Box Cloud Storage... Box | Login

CLEAR ROADS

research for winter highway maintenance

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Home About Research Projects News Resources Qualified Products List (QPL)

Qualified Products List

Due to the administrative restructuring of the Pacific Northwest Snowfighters (PNS) organization, Clear Roads has assumed responsibility of the Qualified Products List. The product testing and administration of the QPL will remain essentially the same along with a few process improvements.

Mission

The mission of the Clear Roads QPL is to “strive to serve the traveling public by evaluating and establishing specifications for products used in winter maintenance that emphasize safety, environmental preservation, infrastructure protection, cost-effectiveness and performance.”

Guidance for Vendors

To begin the process of getting your product approved and listed on the QPL, please complete the [Product Sample Checklist](#) (may need to saved to your desktop prior to completing the document). Once you've completed the checklist, click the yellow submit button at the bottom of page 2. Keep in mind that test results from a third party lab are necessary prior to completing the checklist. Once paperwork is approved, Clear Roads will ask for product samples per the specifications.

The QPL product approval process can be expected to take at least 4 months.

Questions? Please call Patti Caswell at 503-986-3008.

DOCUMENTS

[QPL Listing \(7/29/2020\)](#)

[Chemical Product Protocols \(3/2019\)](#)

QUICK LINKS

[Letter to Vendors](#)

[QPL Contacts](#)

[Lab Photos](#)

Qualified Products List | Clear Roads

QPL_Updated-July-29-2020.pdf

clearroads.org/wp-content/uploads/QPL_Updated-July-29-2020.pdf

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Clear Roads Qualified Product List - PRODUCTS

Date of Listing: July 29, 2020

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Product Name	Manufacturer	Corrosion Rate % Effectiveness	% Concentration	Date Approved	Expiration
Calcium 200	Laird Frendly Chem.	1.4	25%	8/15/2002	12/31/2024
Caliber M1000 AP	Envirotech Services Inc.	25.8	30%	8/25/2004	12/31/2024
Meddown with Shield AP	Envirotech Services Inc.	25.8	30%	8/25/2004	12/31/2024
Hydro-Melt Green	Cargill	25.3	25.0%	8/1/2005	12/31/2024
Meddown APEX with Shield AP	Envirotech Services Inc.	25.1	30%	1/25/2006	12/31/2024
FreezeGuard CI Plus	Compass Minerals	12.2	30%	8/28/2009	12/31/2024
Ice Whone 114F	Sarna Ecological Appl.	28.6	25%	8/9/2007	12/31/2024
FreezeGuard LTLE CI Plus	Compass Minerals	12.3	27%	8/13/2011	12/31/2024
HydroMelt Liquid Deicer	Cargill	28	25.0%	8/16/2011	12/31/2024
FreezeGuard CI Plus Sub Zero	Compass Minerals	14.1	27.5%	10/11/2011	12/31/2024
Torch IB	GMCO Corporation	25.3	25.0%	1/10/2013	12/31/2024
Torch CI	GMCO Corporation	21.2	30.0%	1/10/2013	12/31/2024
Meddown Apex	Envirotech Services Inc.	22.4	30.0%	4/16/2014	12/31/2024
Meddown Inhibited	Envirotech Services Inc.	14.1	30.0%	4/28/2014	12/31/2024
ProMelt MAG 30 INH	Innovative Surface Solutions	25.2	30.0%	7/31/2015	12/31/2024
ProMelt Ultra 1000 INH	Innovative Surface Solutions	25.2	27.0%	7/31/2015	12/31/2024
Torch LT	GMCO Corporation	22	30.0%	12/11/2015	12/31/2024
NextGen Liquid Deicer	Paradigm Group	25	30.0%	6/12/2017	12/31/2024

Note-Iceban 200 was formerly Iceban Performance Plus M.
Torch IB was formerly Ice Ban 305, Torch CI was FreezeGuard 0 CCI, Torch LT was NextGen Torch.
Those products marked with an asterisk (*) indicates that the stratification can be seen and agitation is required.

Product Name	Manufacturer	Corrosion Rate % Effectiveness	% Concentration	Date Approved	Expiration
Liquid Snow Armor	Dow Chemical	28	30%	8/25/1999	12/31/2024
Winter Triage 01	Kelp Technologies	16.5	32%	8/13/1999	12/31/2024
Guardian YG	Tiger Calcium Services	27.7	29%	1/8/2001	12/31/2024
Road Guard Plus	Tiger Calcium Services	18	25%	6/5/2006	12/31/2024
Calcium chloride with Boost (CCB)	AgriChem West	18.4	32%	4/10/2014	12/31/2024
Meddown Apex-C	Envirotech Services Inc.	24.5	29%	8/21/2015	12/31/2024
CI1000 Pro	Envirotech Services Inc.	28	27%	9/8/2015	12/31/2024

Product Name	Manufacturer	Corrosion Rate % Effectiveness	% Concentration	Date Approved	Expiration
Liquid CMA 22%	Cryotech	11.1	25%	8/18/1999	12/31/2024
EC-CMA 52%	Sure Crop Farm Services	2.8	25%	8/13/1999	12/31/2024

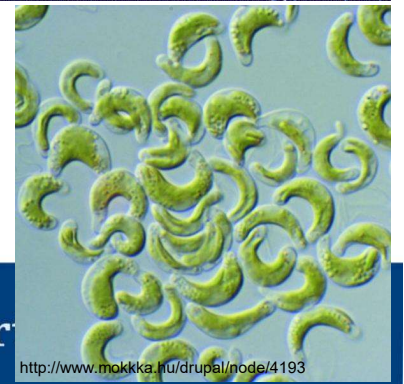
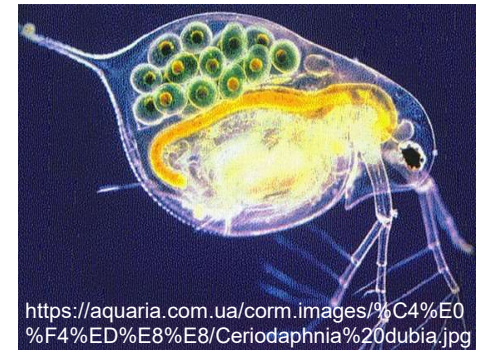
What does it take to be on QPL?

- Elemental Analysis

Arsenic	5.0
Barium	100.0
Cadmium	0.20
Chromium	1.0
Copper	1.0
Lead	1.0
Mercury	0.05
Selenium	5.0
Zinc	10.00
Phosphorus	2500.
Cyanide	0.20

- Other Testing Required

Ammonia - Nitrogen
Total Kjeldahl Nitrogen
Nitrate and Nitrite - Nitrogen
Biological Oxygen Demand
Chemical Oxygen Demand
Frictional Analysis
Toxicity Testing
 Rainbow Trout or Fathead Minnow Toxicity Test
 Ceriodaphnia Dubia Reproductive and Survival Bioassay
 Selenastrum Capricornutum Algal Growth



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Where do these numbers come from?

- Your State
- The Environmental Protection Agency (EPA)
 - <https://www.epa.gov/wqc/national-recommended-water-quality-criteria-aquatic-life-criteria-table> (aquatic life)
 - <https://www.epa.gov/wqc/national-recommended-water-quality-criteria-human-health-criteria-table> (human health)



Let's talk salt

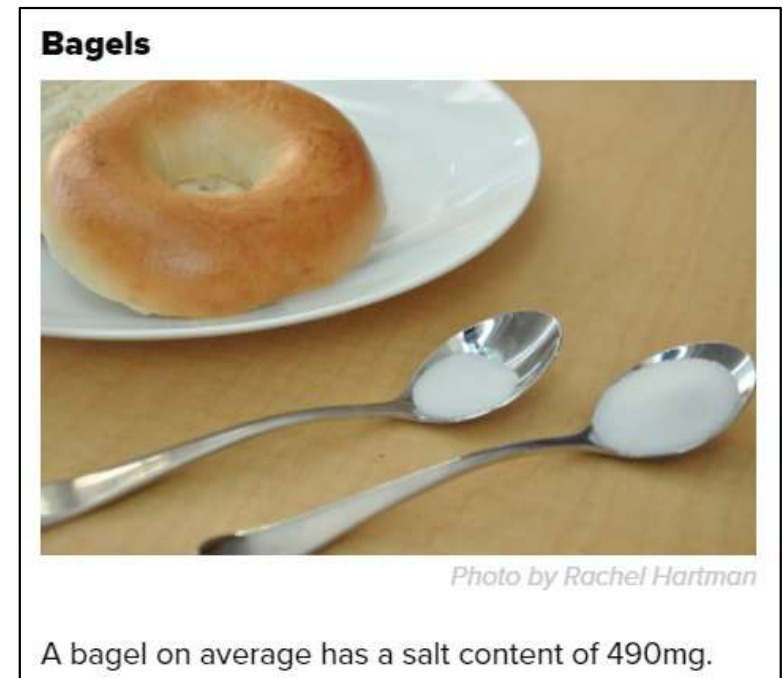
- EPA water quality standard (*secondary drinking water standard)

$$250 \text{ mg/L} = \begin{cases} 250 \text{ mg Chloride} \\ 1 \text{ Liter water} \end{cases}$$

*Maximum contaminant level, causes undesirable taste or odor, undesirable effects to the body, damage to equipment....

What 250 mg/L of salt looks like!

- 250 mg = about 1/10 of a teaspoon of salt



<https://spoonuniversity.com/lifestyle/salt-content-favorite-foods-actually-looks-like>

A little chemistry

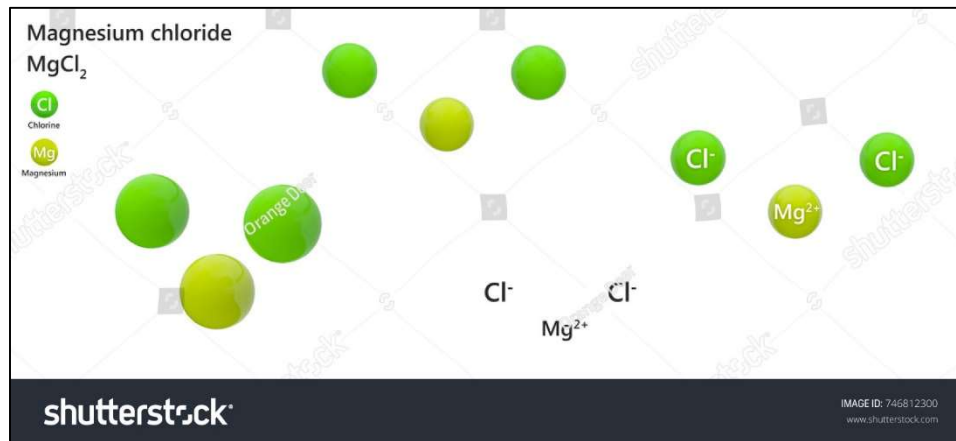
NaCl



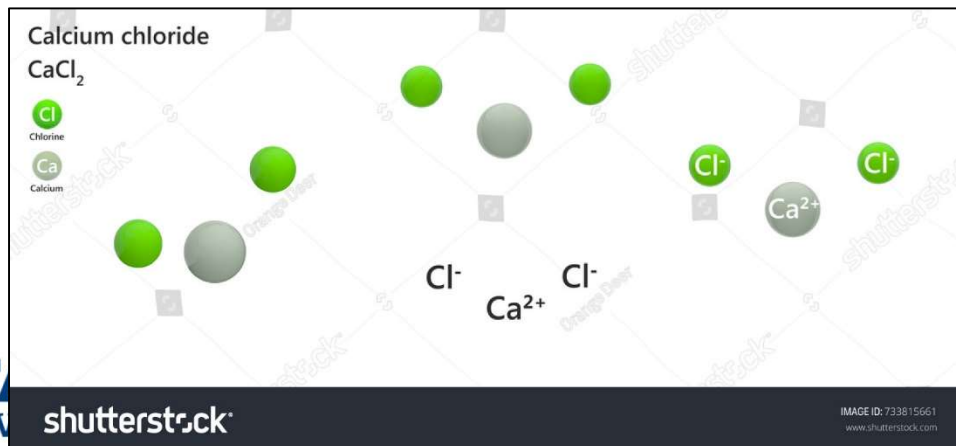
Or salt is 40% Sodium and 60% Chloride!

[100 g of NaCl = 39.34 g Na + 60.66 g Cl.]

MgCl₂



CaCl₂



Both Mag and Calcium chloride have twice as much chloride as NaCl

What does this have to do with winter maintenance?

- Chloride (drinking water standard):
 - 250 mg/L
- Chloride (aquatic life standards):
 - 230 mg/L Chronic (longer term exposure)
 - 860 mg/L Acute (1 time exposure)



Products used by WSDOT and Application Rates

- **NaCl** (32 to 15°F)
 - Solid – 100 to 800 lbs/l-m
 - Liquid – 10 to 40 gal/l-m
 - Pre-wet – 8 to 20 gal/l-m
- **MgCl₂** (32 to -5°F) and **CaCl₂** (32 to -15°F)
 - Solid – 100 to 500 lbs/l-m
 - Liquid – 10 to 40 gal/l-m
 - Pre-wet – 8 to 20 gal/l-m

Other products?

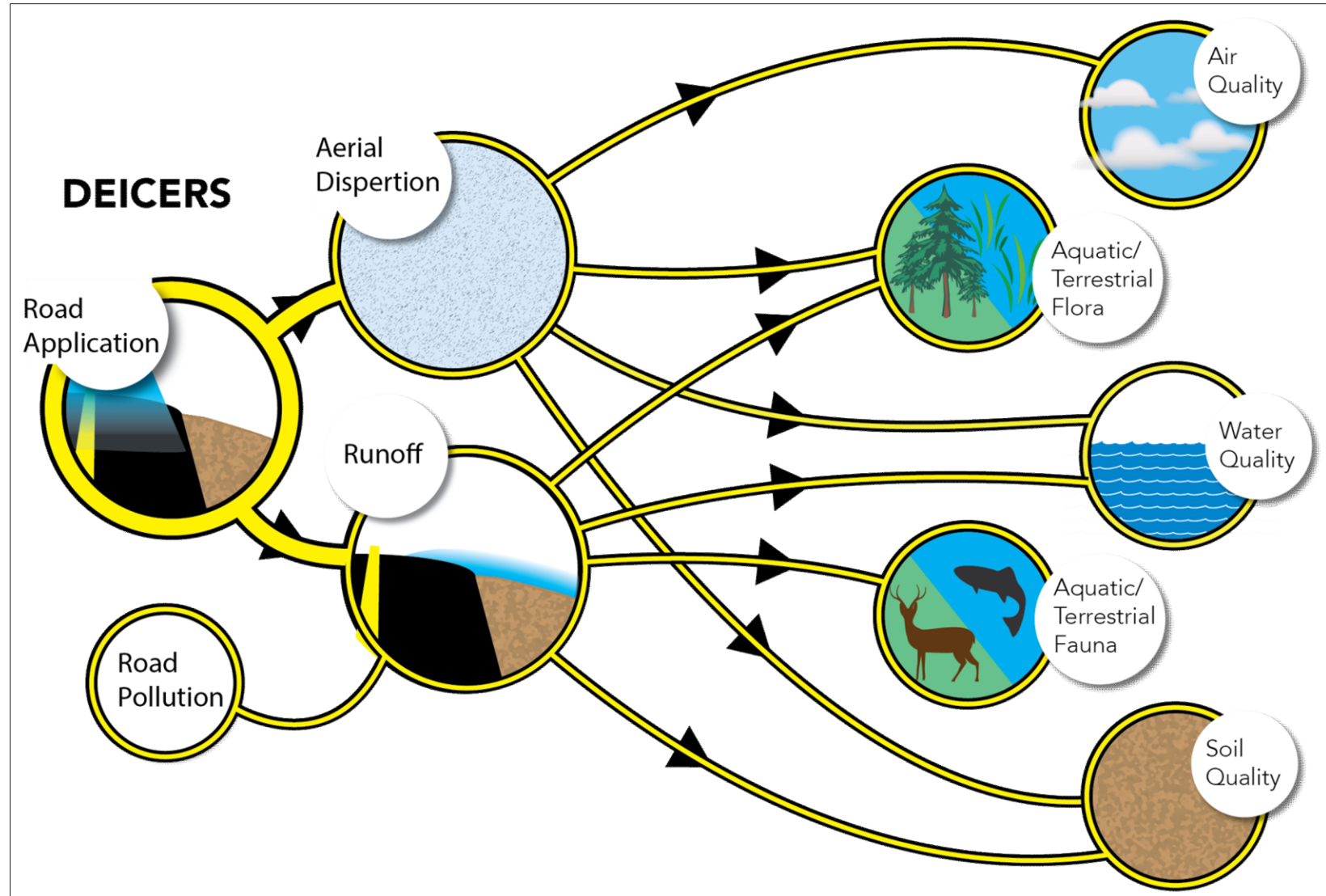
What application rates do you use?

For anti-icing?

For pre-wetting?

For deicing?

Once applied, where do deicers go?



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Let's play a game (not really just math)



<u>Pounds (lbs)</u>	<u>Milligrams (mg)</u>	<u>Pond [NaCl]</u>
50	22,680,000	11 mg/L
100	45,360,000	22 mg/L
250	113,400,000	57 mg/L
500	226,800,000	113 mg/L
1000	453,600,000	227 mg/L

So where do we go from here?

1. What are actual water chloride levels?
2. And what toxicological impacts are being observed?

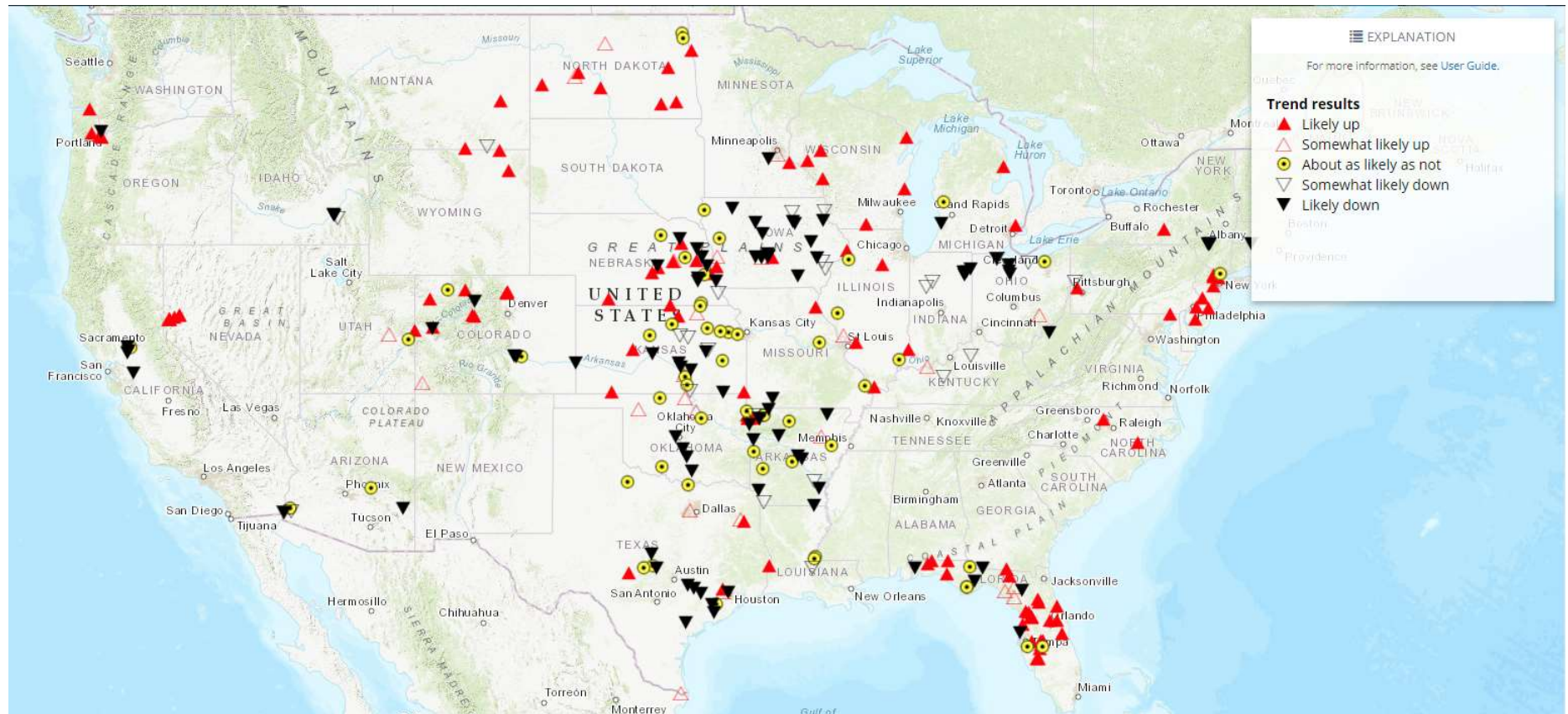


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USGS Stream and Surface Water Chloride Concentrations from 2002-2012



<https://nawqatrends.wim.usgs.gov/swtrends/>

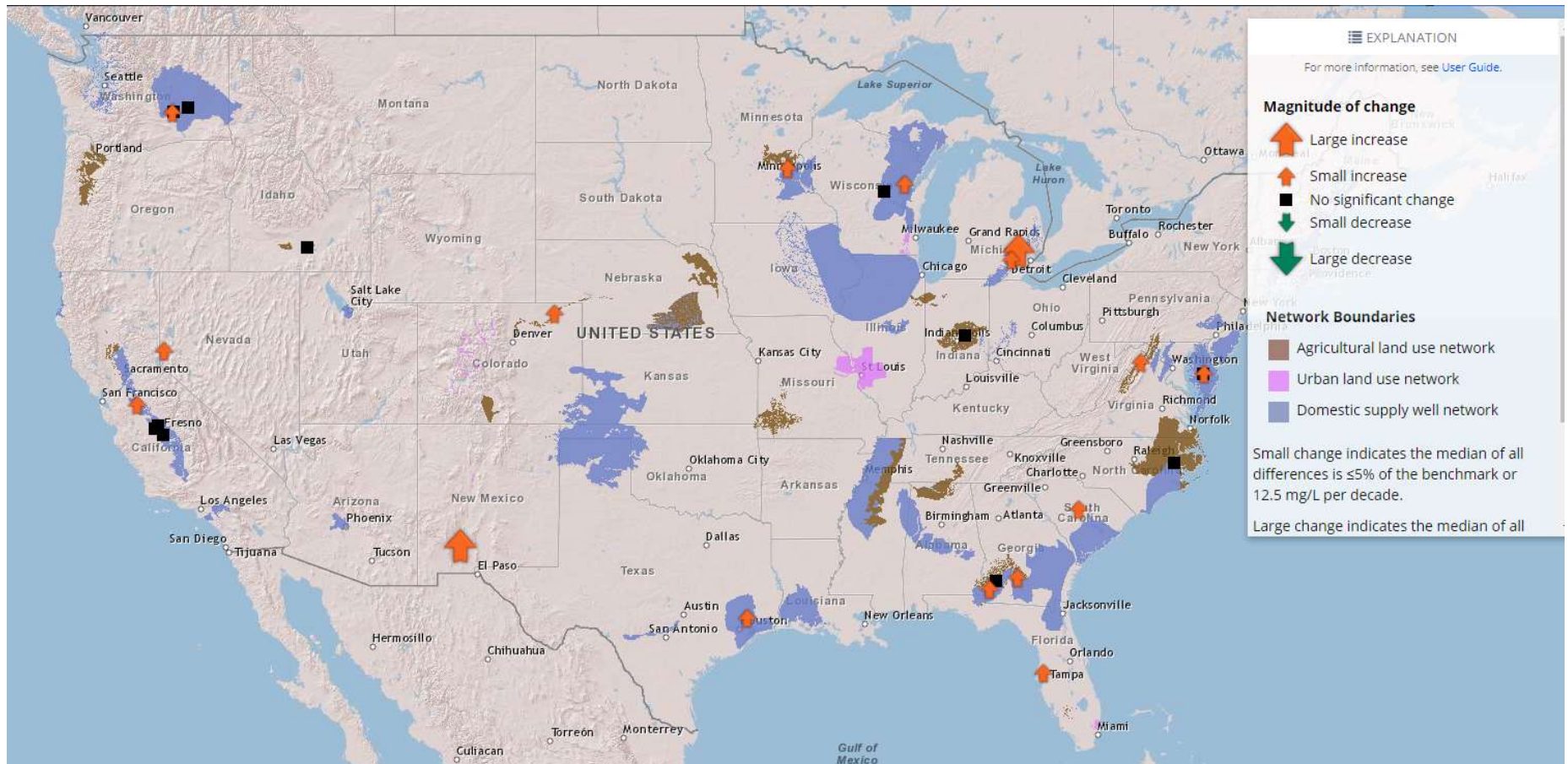


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USGS Groundwater Chloride Concentration changes from 1988-2016



<https://nawqatrends.wim.usgs.gov/Decadal/>

Chloride numbers from the field

- USGS (Corsi et al., 2014)
 - 29% of the sites exceeded the EPA (230 mg/L)
 - by an average of more than 100 days per year from 2006 - 2011, almost double the amount of days from 1990 -1994.
 - The lowest chloride concentrations were in watersheds that had little urban land use or cities without much snowfall.

<https://www.usgs.gov/news/urban-stream-contamination-increasing-rapidly-due-road-salt>



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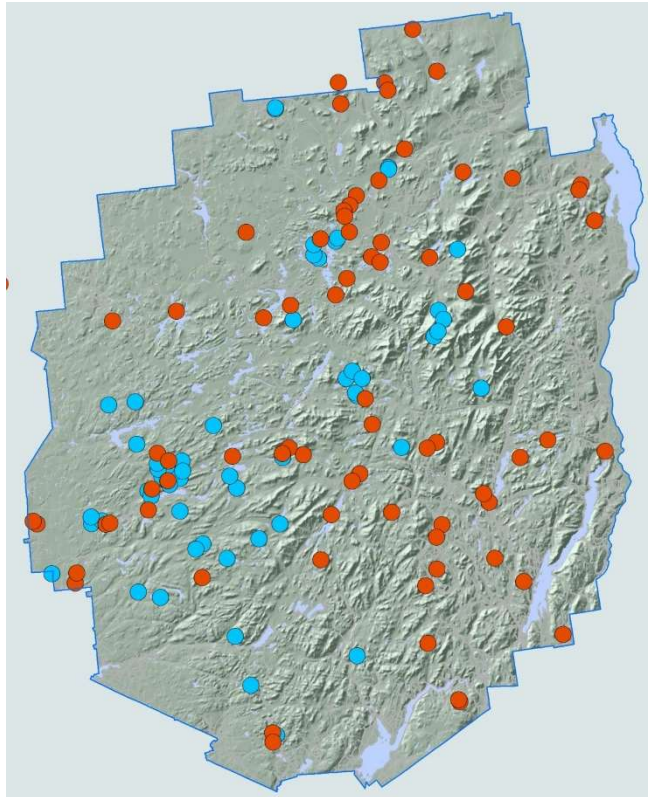
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USGS (Corsi et al., 2014)

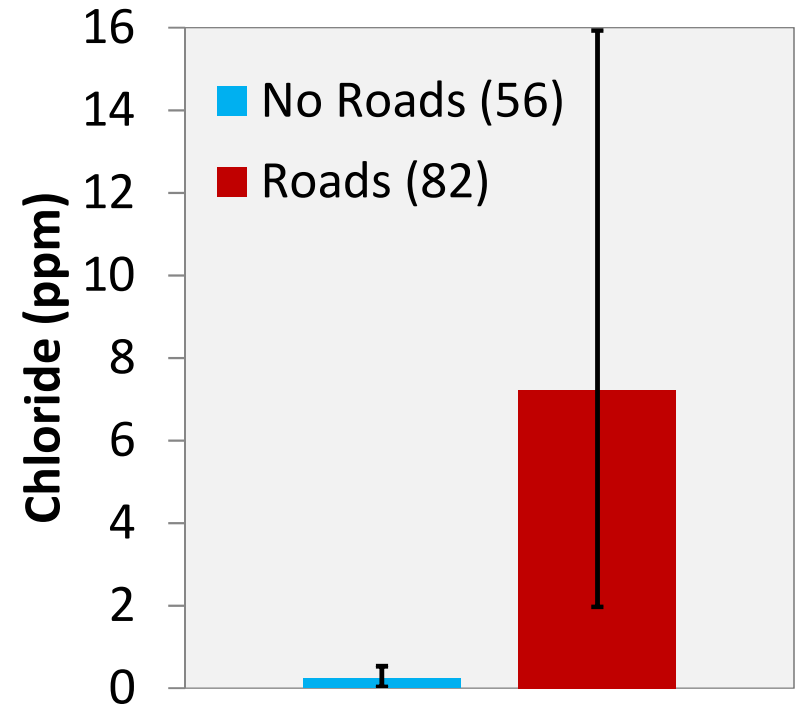
- In 16 of the streams, winter chloride concentrations increased over the study period.
- In 13 of the streams, chloride concentrations increased over the study period during non-deicing periods such as summer.
 - chloride infiltrating the groundwater system during the winter, then slowly released to the streams throughout the year.



Regional Salinization caused by WMO



Kelting, D. L., Laxson, C. L., & Yerger, E. C. (2012). Regional analysis of the effect of paved roads on sodium and chloride in lakes. *Water Research*, 46(8), 2749-2758.



- <0.5ppm w/**no roads**
- 14X higher w/**roads**



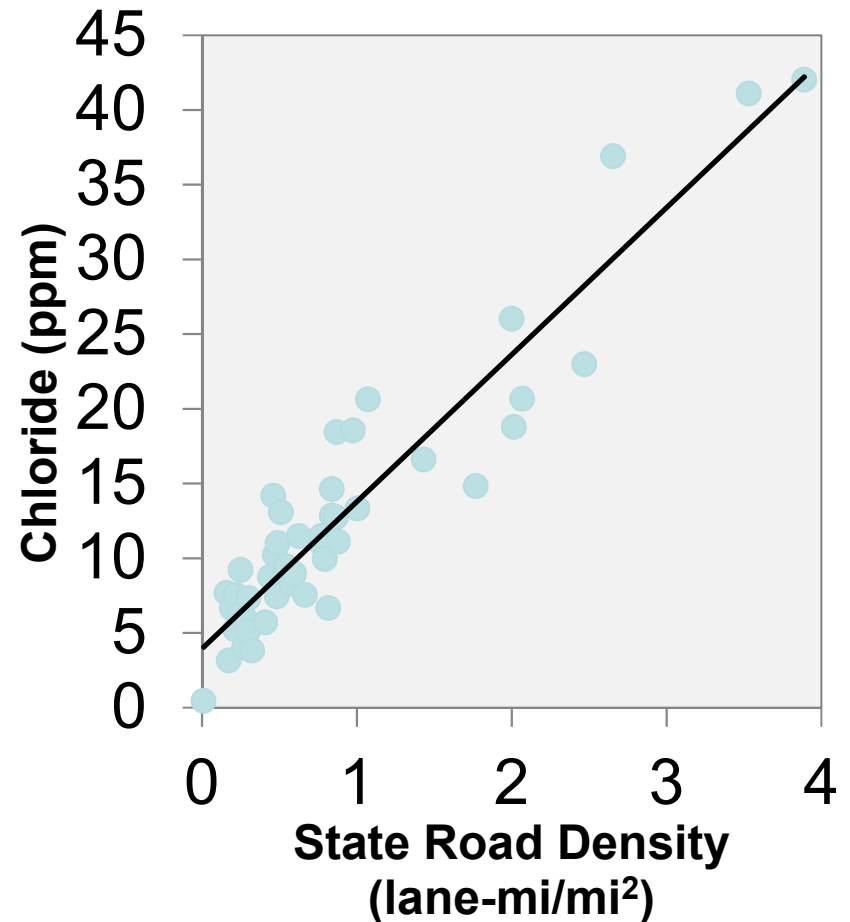
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Lake Chloride Concentrations tied to State Road Density

- Higher state road density equals higher salt load
- No relationship found between local roads and chloride concentration.



Kelting, D. L., Laxson, C. L., & Yerger, E. C. (2012). Regional analysis of the effect of paved roads on sodium and chloride in lakes. *Water Research*, 46(8), 2749-2758.



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It is not just surface water; it is ground water too...

- They found the UNEXPECTED
 - Think slow, glacial time scale..
1. Long term increases in concentration
 2. High [Cl⁻] in summer
 3. Higher [Cl⁻] downstream in summer

Stuart Findlay and Vicky Kelly (Cary Institute, 2018)



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Stuart Findlay and Vicky Kelly (Cary Institute, 2018)

- Background < 10 mg/L
 - Environmental effects (sub lethal) ~ 100 mg/L
 - Lethal > 1000 mg/L
-

- EPA Drinking Water Std. 250 mg/L
- EPA chronic 230 mg/L
- EPA acute 860 mg/L



Summary of Reported Chloride concentrations

Freshwater Ecosystem	Range due to natural sources (mg/L)	Range from road salt contamination
Lakes	0 - 10	6 – 1,000
River/Stream	1 - 20	10 – 7,730
Wetland/Pond	0 - 12	10 – 13,500

Sources of chloride

- 80% from deicing (DOT, local, private)
- 5-10% from water softeners

(Findlay and Kelly, Cary Institute, 2018; Hintz and Relyea, 2018)



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The Ecological Perspective

- Salt negatively affects species across all trophic levels – biofilms to fish
- Salt impacts vary by concentration and species
- Typical impacts for species were not death, but reduction in growth and reproduction
- Community level impacts were reductions in biodiversity, and encouragement of salt tolerant species

Hintz, W.D. and Relyea, R.A. (2018) A review of the species, community, and ecosystem impacts of road salinization in fresh waters. *Freshwater Biology*.



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Other documented impacts of Deicers

- Deicers can cause...
 - Mobilizations of heavy metals
 - Impacts to or death of aquatic & terrestrial species
 - Loss of native species => increase in invasive species (aquatic & terrestrial)
 - Wildlife-vehicle collisions



<https://rachelcarsoncouncil.org/safe-deicing-products/>



<https://www.hoosier-aaa.com/insurance/wildlife-animal-vehicle-collisions>

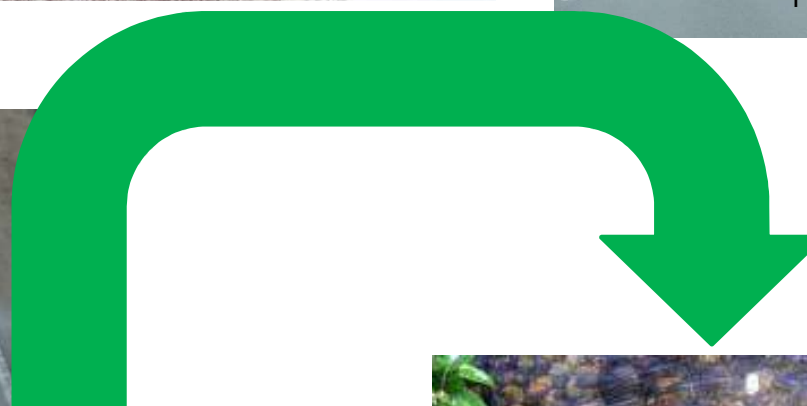


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Impacts of Salt and Chloride Based Deicers



Impacts of Sand and Abrasives



BOD

(Biological/Biochemical Oxygen Demand)

- “The amount of dissolved oxygen needed by aerobic biological organisms to break down material in water at a specific temperature or a specific time.”

https://en.wikipedia.org/wiki/Biochemical_oxygen_demand

<https://www.slideshare.net/jamesmacroony/biochemical-oxygen-demand>

	BOD Standards (mg/L)
Most pristine rivers	< 1
Moderately polluted rivers	2 - 8
Ordinary domestic sewage	150 - 200
Treated sewage	< 20

BOD Level in mg/liter	Water Quality
1 - 2	Very Good: There will not be much organic matter present in the water supply.
3 - 5	Fair: Moderately Clean
6 - 9	Poor: Somewhat Polluted - Usually indicates that organic matter present and microorganisms are decomposing that waste.
100 or more	Very Poor: Very Polluted - Contains organic matter.

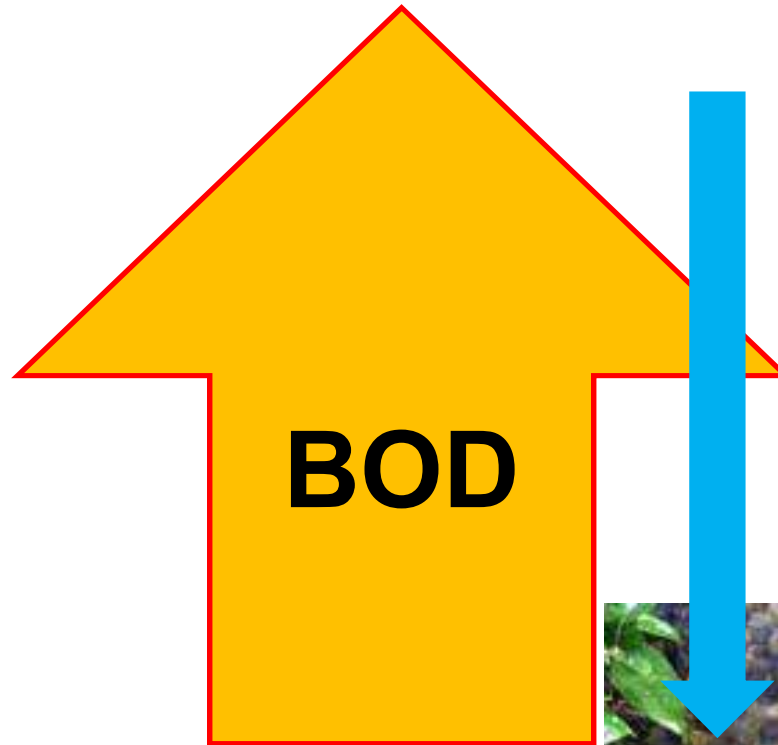
Ag-based, Acetate, Formates & Glycols

Benefits

- Break down in the environ.
- Less corrosive than chlorides

Not so good

- Higher costs
- Exert a higher BOD



Reduces available oxygen for organisms in the soil and aquatic environments

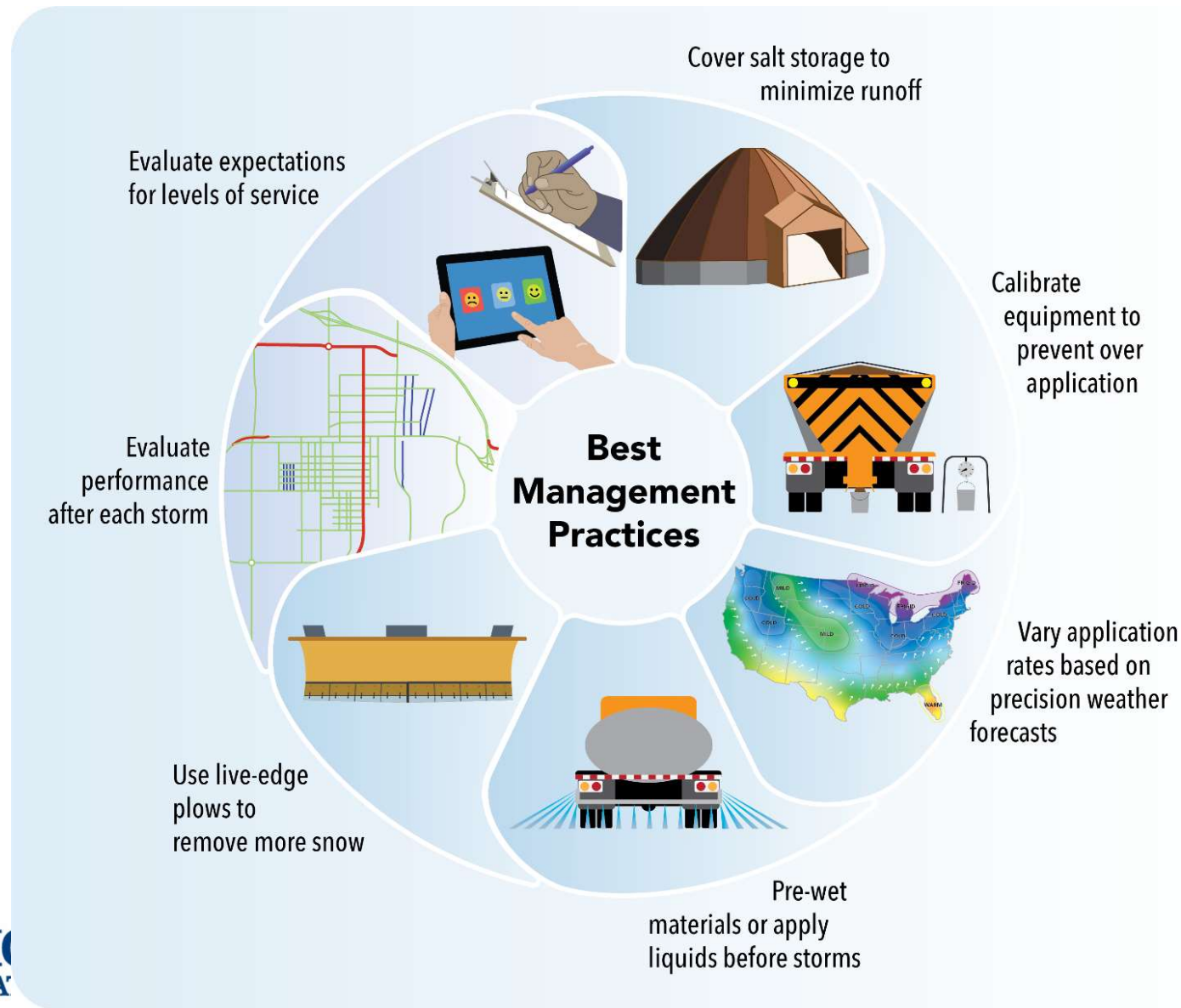


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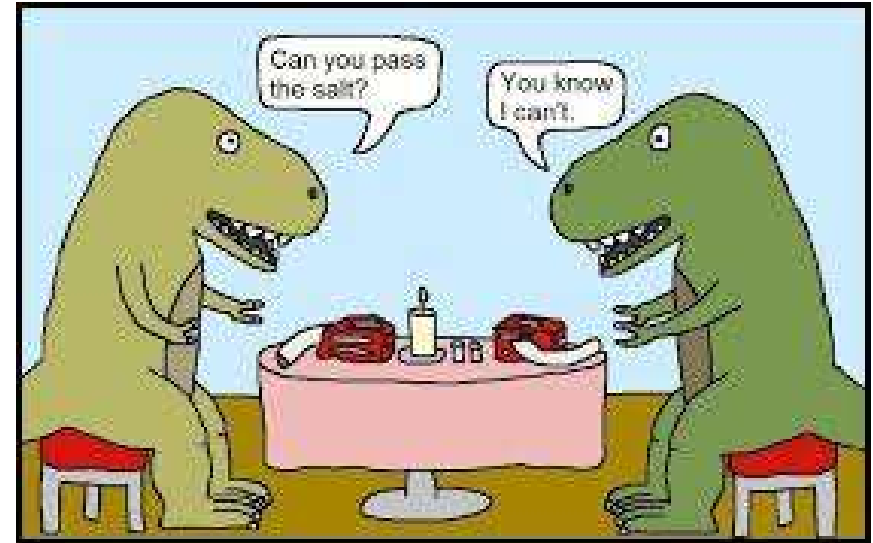
Where does this leave us...



Where does this leave us...

<https://encrypted-tbn0.gstatic.com/images?q=tbn%3AA-Nd9GcR5MGKmoiGc9aoBATLUIXWF-IOS-RHeJ9vfu4w&usqp=CAU>

- Invent a better deicer
- Invest in a different deicing system



Heated pavements?

Resources

- www.clearroads.org



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Questions?

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<https://westerntransportationinstitute.org/>



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